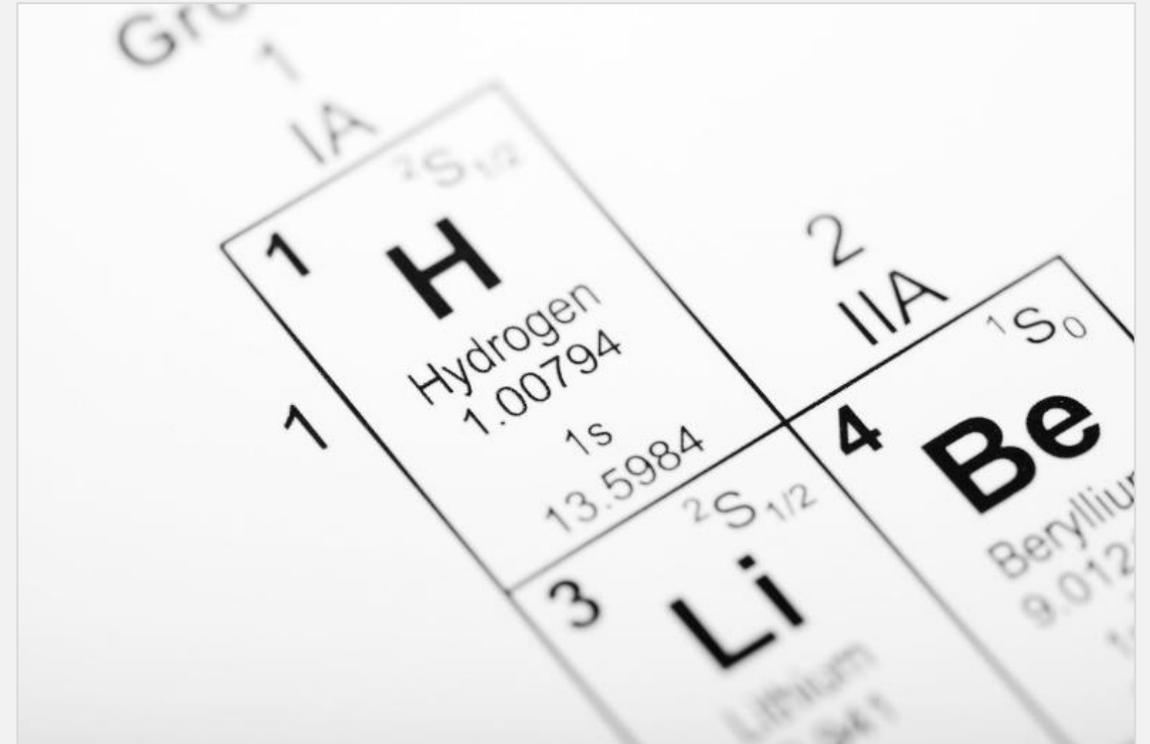


Outline

- 1 What is Hydrogen (H₂)?
- 2 What is the current role of H₂?
- 3 Why are we talking so much about H₂?
- 4 What is the H₂ economy?
 - H₂ production
 - H₂ storage
 - H₂ transport & distribution
 - H₂ utilization
- 5 What is the role of H₂ in a carbon neutral economy (European case)?
- 6 What is Portugal planning for H₂?



What is Hydrogen?

- > The most abundant chemical substance in the Universe.
- > The lightest element in the periodic table.
- > Contains more energy per unit of mass than natural gas or gasoline (3X) – lower energy per volume (1/10 of natural gas)

larger volumes of hydrogen are needed to meet identical energy demands as compared with other fuels

Hydrogen

atomic number: 1
 symbol: H
 electron configuration: 1s¹
 name: hydrogen
 atomic weight: 1.008
 acid-base properties of higher-valence oxides
 crystal structure: hexagonal
 physical state at 20 °C (68 °F): Gas

Other nonmetals (orange square)
 Gas (dotted line)
 Hexagonal (hexagon symbol)
 Equal relative strength (circle with horizontal line)

© Encyclopædia Britannica, Inc.

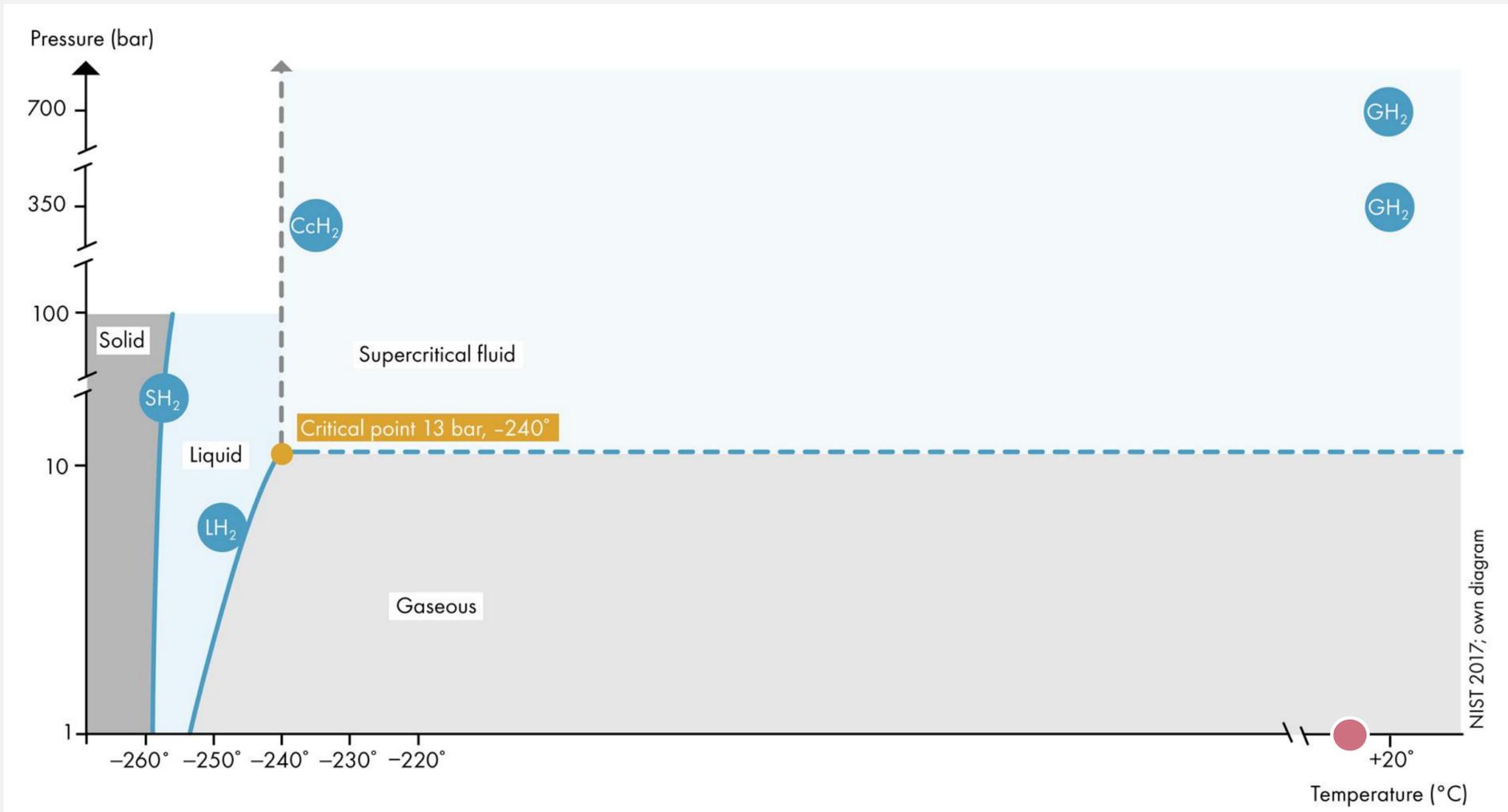
Physical properties of hydrogen

Property	Hydrogen	Comparison
Density (gaseous)	0.089 kg/m ³ (0°C, 1 bar)	1/10 of natural gas
Density (liquid)	70.79 kg/m ³ (-253°C, 1 bar)	1/6 of natural gas
Boiling point	-252.76°C (1 bar)	90°C below LNG
Energy per unit of mass (LHV)	120.1 MJ/kg	3x that of gasoline
Energy density (ambient cond., LHV)	0.01 MJ/L	1/3 of natural gas
Specific energy (liquefied, LHV)	8.5 MJ/L	1/3 of LNG
Flame velocity	346 cm/s	8x methane
Ignition range	4–77% in air by volume	6x wider than methane
Autoignition temperature	585°C	220°C for gasoline
Ignition energy	0.02 MJ	1/10 of methane

Notes: cm/s = centimetre per second; kg/m³ = kilograms per cubic metre; LHV = lower heating value; MJ = megajoule; MJ/kg = megajoules per kilogram; MJ/L = megajoules per litre.

Source: IEA, 2019a

Phase H₂ diagram



Source: Shell, 2017

The early entrance of H₂ in the energy system

- > In the beginning of the XIX century H₂ was incorporated in street lighting in Europe and USA as town gas (produced through coal gasification)
- > H₂ was around 50% of town gas
- > Why town gas?
 - > Economic: Cheaper than whale oil
 - > Quality of services: Brighter and safer flame

- Widespread adoption of town gas in UK around 1820
- In Lisbon the public lighting through town gas started in 1848 (Chiado)



- > When electricity (and in some countries natural gas) appeared, town gas started to disappear
- > In Portugal for example the first electric lighting appeared in 1878 and the town gas continued in the streets of Lisbon up to 1965 (Bairro Alto and Santa Catarina).

Is H₂ safe?

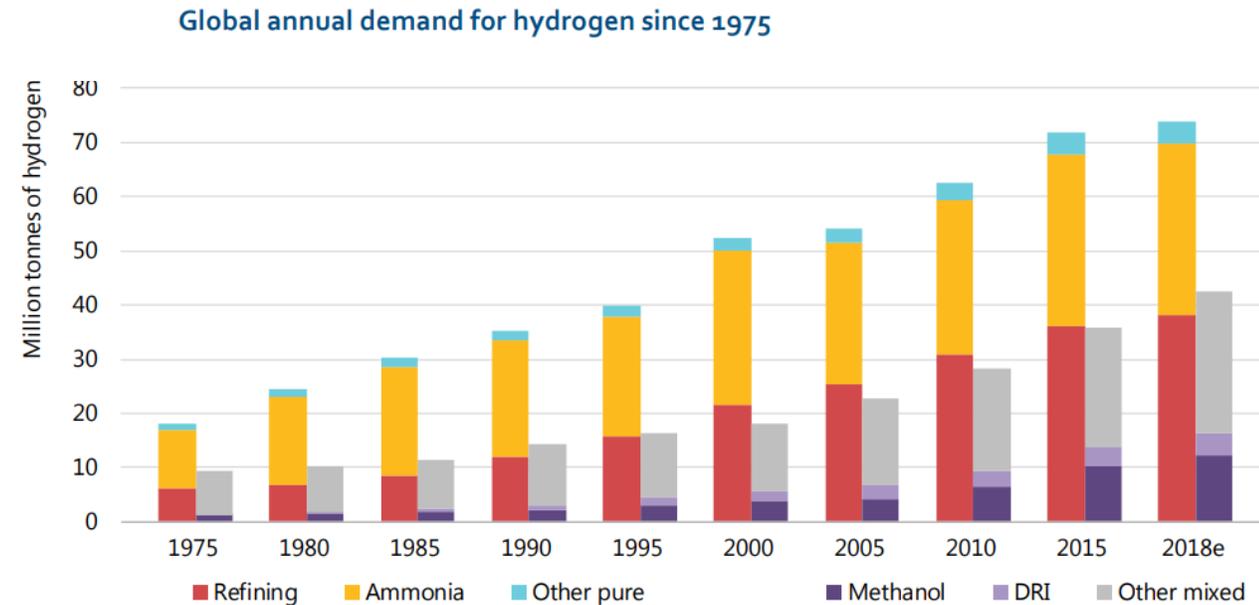
1. H₂ is **not toxic**, unlike conventional fuels.
2. H₂ is **14X lighter than air and 57X lighter** than gasoline vapor. When released, hydrogen will typically rise and disperse rapidly, reducing the risk of ignition at ground level..
3. H₂ has a **lower radiant heat than conventional gasoline**, meaning the air around the flame of hydrogen is not as hot as around a gasoline flame.
4. H₂ has a **higher oxygen requirement for explosion than fossil fuels**. It can be explosive with oxygen concentrations between 18-59% while gasoline can be explosive at oxygen concentrations between 1-3%.



Hindenburg disaster
1937



What is the current role of H₂?



Notes: DRI = direct reduced iron steel production. Refining, ammonia and "other pure" represent demand for specific applications that require hydrogen with only small levels of additives or contaminants tolerated. Methanol, DRI and "other mixed" represent demand for applications that use hydrogen as part of a mixture of gases, such as synthesis gas, for fuel or feedstock.

Source: IEA 2019. All rights reserved.

Around 70 MtH₂/yr is used today in pure form, mostly for oil refining and ammonia manufacture for fertilisers; a further 45 MtH₂ is used in industry without prior separation from other gases.

Current H₂ uses:

- > refining petroleum (e.g., lower the sulfur content of diesel fuel),
- > producing fertilizer (ammonia)

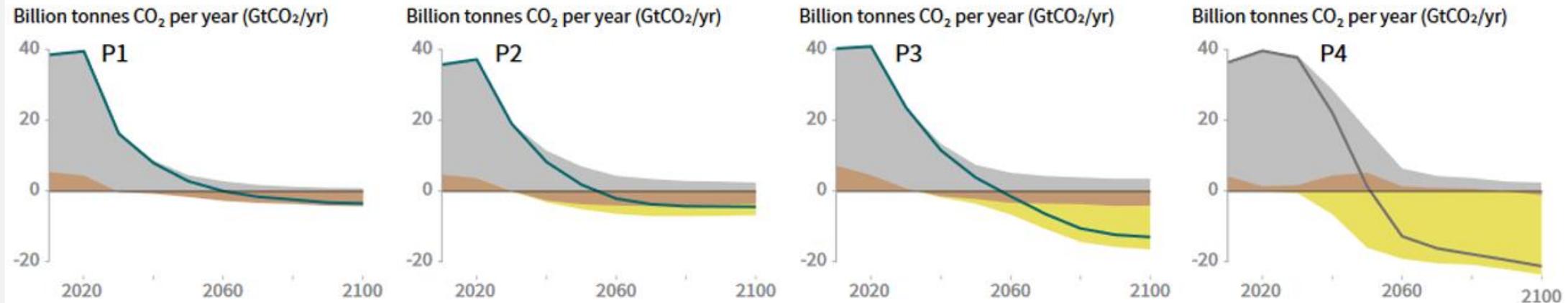
Source: IEAa, 2019

Why are we talking so much about H₂?

Climate Change -
Limiting global warming to 1.5°C compared to pre-industrial levels

Breakdown of contributions to global net CO₂ emissions in four illustrative model pathways

● Fossil fuel and industry ● AFOLU ● BECCS



P1: A scenario in which social, business and technological innovations result in lower energy demand up to 2050 while living standards rise, especially in the global South. A downsized energy system enables rapid decarbonization of energy supply. Afforestation is the only CDR option considered; neither fossil fuels with CCS nor BECCS are used.

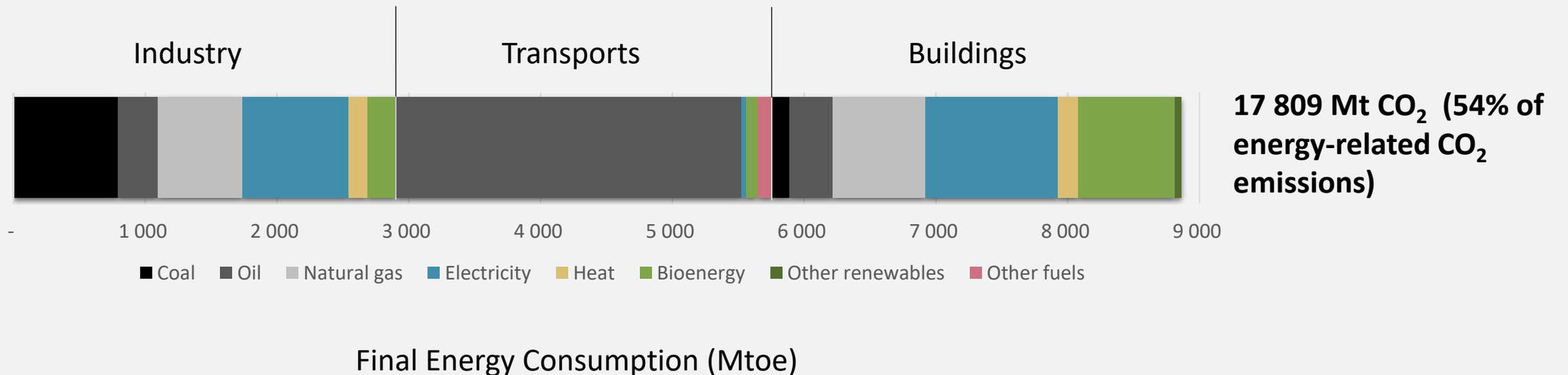
P2: A scenario with a broad focus on sustainability including energy intensity, human development, economic convergence and international cooperation, as well as shifts towards sustainable and healthy consumption patterns, low-carbon technology innovation, and well-managed land systems with limited societal acceptability for BECCS.

P3: A middle-of-the-road scenario in which societal as well as technological development follows historical patterns. Emissions reductions are mainly achieved by changing the way in which energy and products are produced, and to a lesser degree by reductions in demand.

P4: A resource- and energy-intensive scenario in which economic growth and globalization lead to widespread adoption of greenhouse-gas-intensive lifestyles, including high demand for transportation fuels and livestock products. Emissions reductions are mainly achieved through technological means, making strong use of CDR through the deployment of BECCS.

Why are we talking so much about H₂?

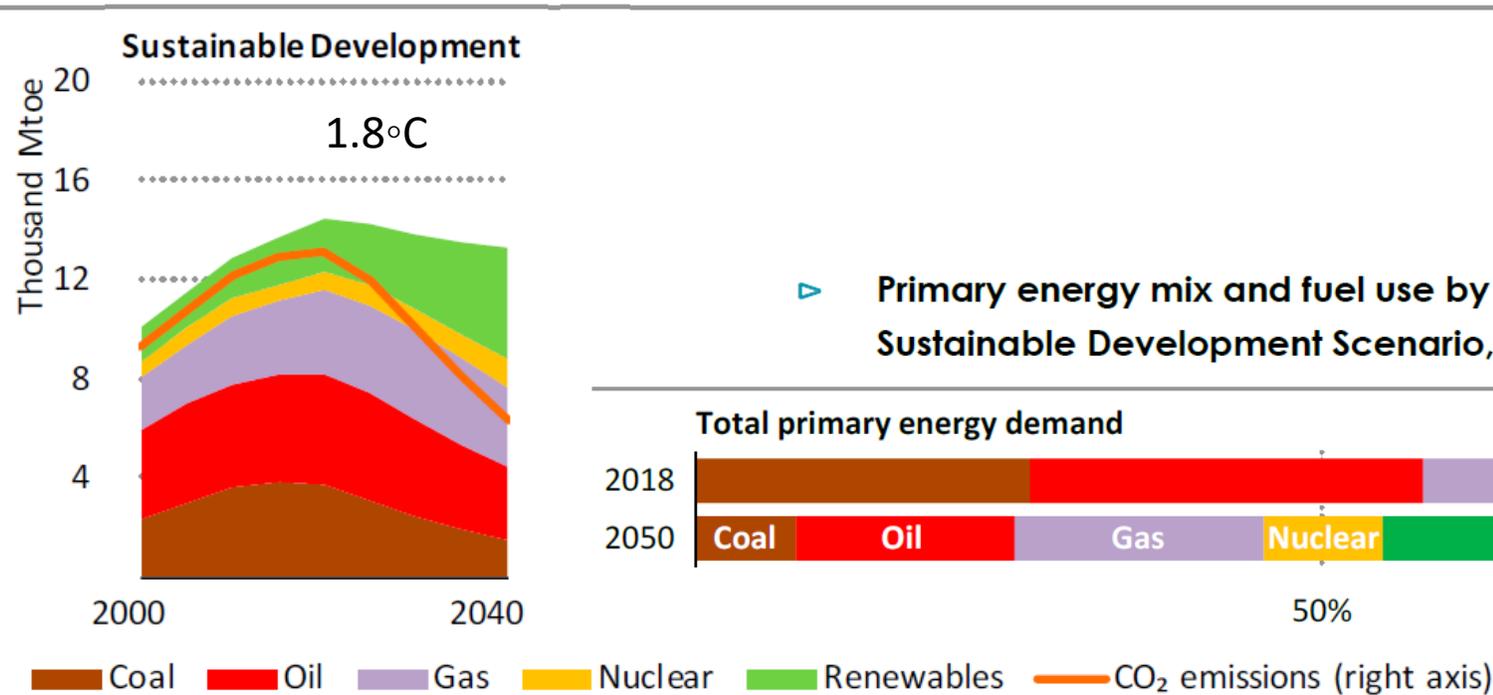
- > H₂ can be an alternative energy vector to lower the carbon intensity of transport (and heating)



Source: IEAb, 2019

Why are we talking so much about H₂?

World primary energy demand by fuel and related CO₂ emissions by scenario

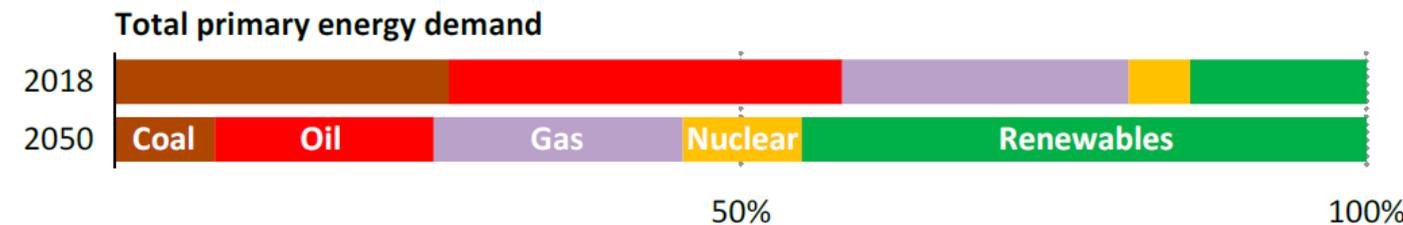


Existing policies and announced targets slow growth in global emissions to 2040, but they are not strong enough to force a peak in an expanding energy system

Energy system transformation

- > Renewable energy sources (RES) will have a major contribution in reduction GHG
- > Most of RES are intermittent

Primary energy mix and fuel use by sector in the Sustainable Development Scenario, 2018 and 2050

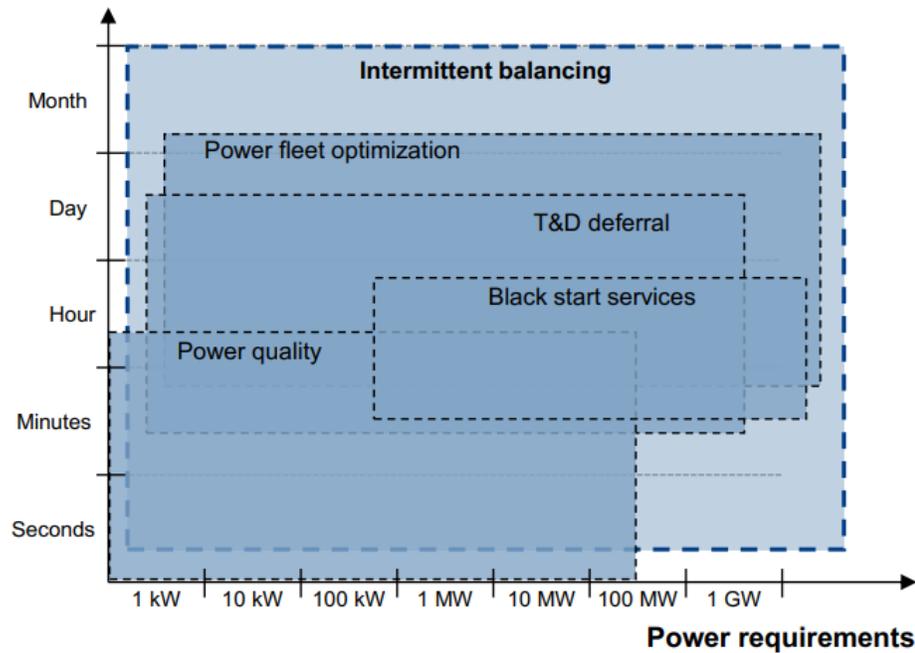


Source: IEAb, 2019

Why are we talking so much about H₂?

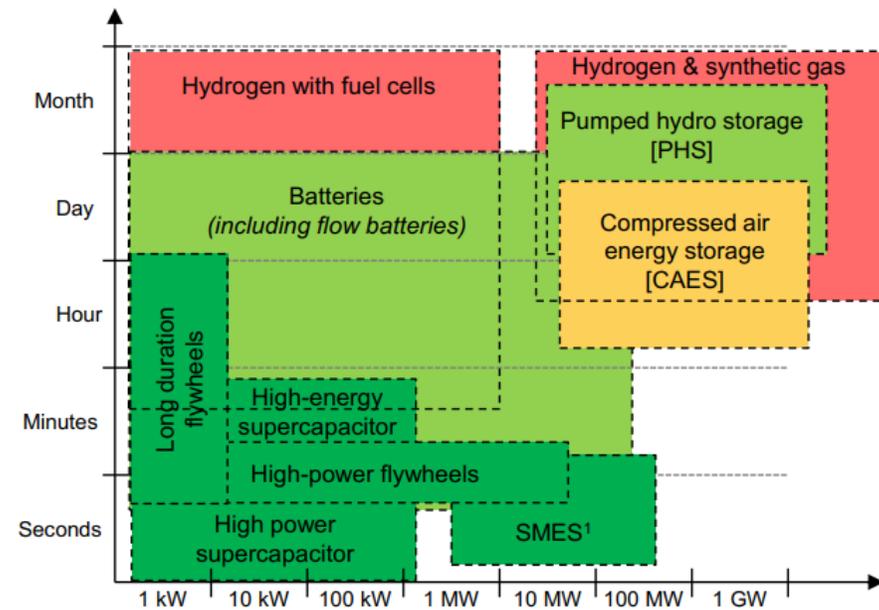
Storage Applications Requirements² Discharge time vs. power requirements (MW)

Discharge Time



Electricity storage technologies' features Discharge time vs. power requirements (MW)

Discharge Time



Electricity Storage

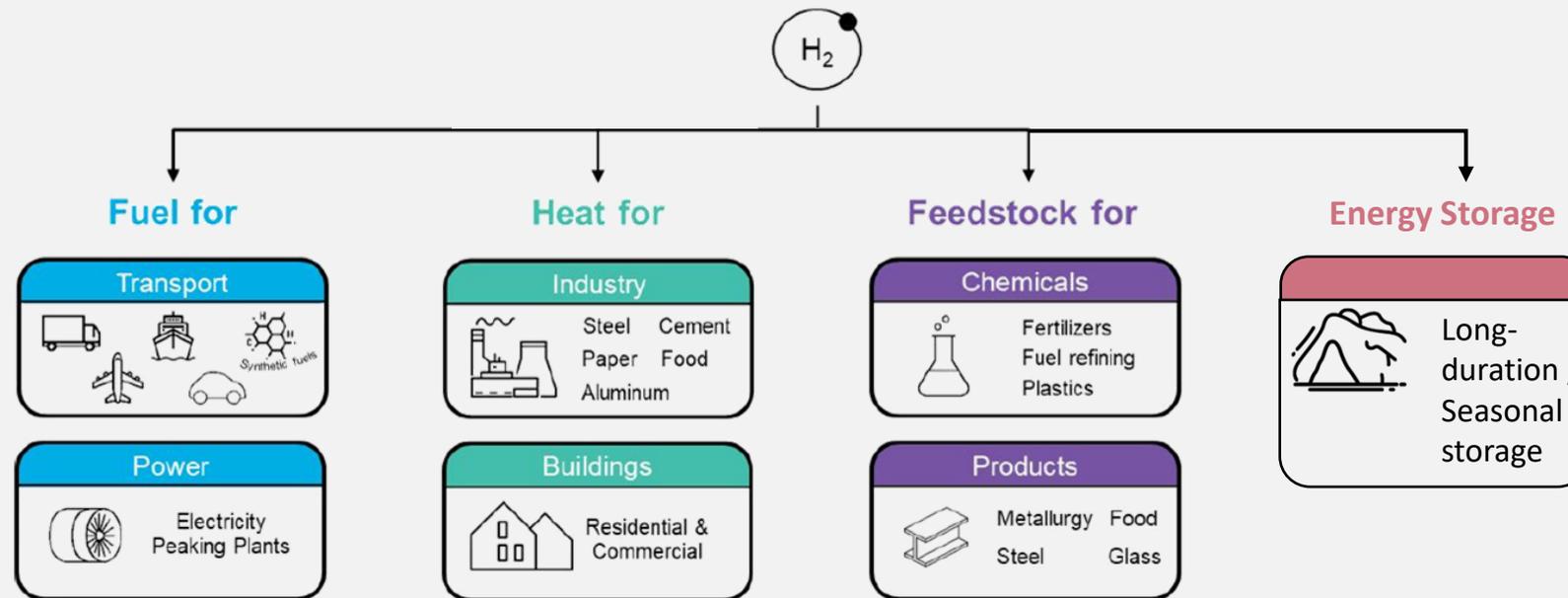
- > H₂ can be storage at large scale
- > Allows a seasonal storage

1. SMES: superconducting magnetic energy storage; 2. For more information on storage applications, please refer to the Hydrogen FactBook; 3. T&D for transmission & distribution
Source: A.T. Kearney Energy Transition Institute based on US DoE (2011), "Energy Storage Program Planning Document".

Hydrogen-based energy conversion

Source: AT Kearney Energy Transition Institute, 2014.

Why are we talking so much about H₂?



Source: Adapted from Bloomberg NEF

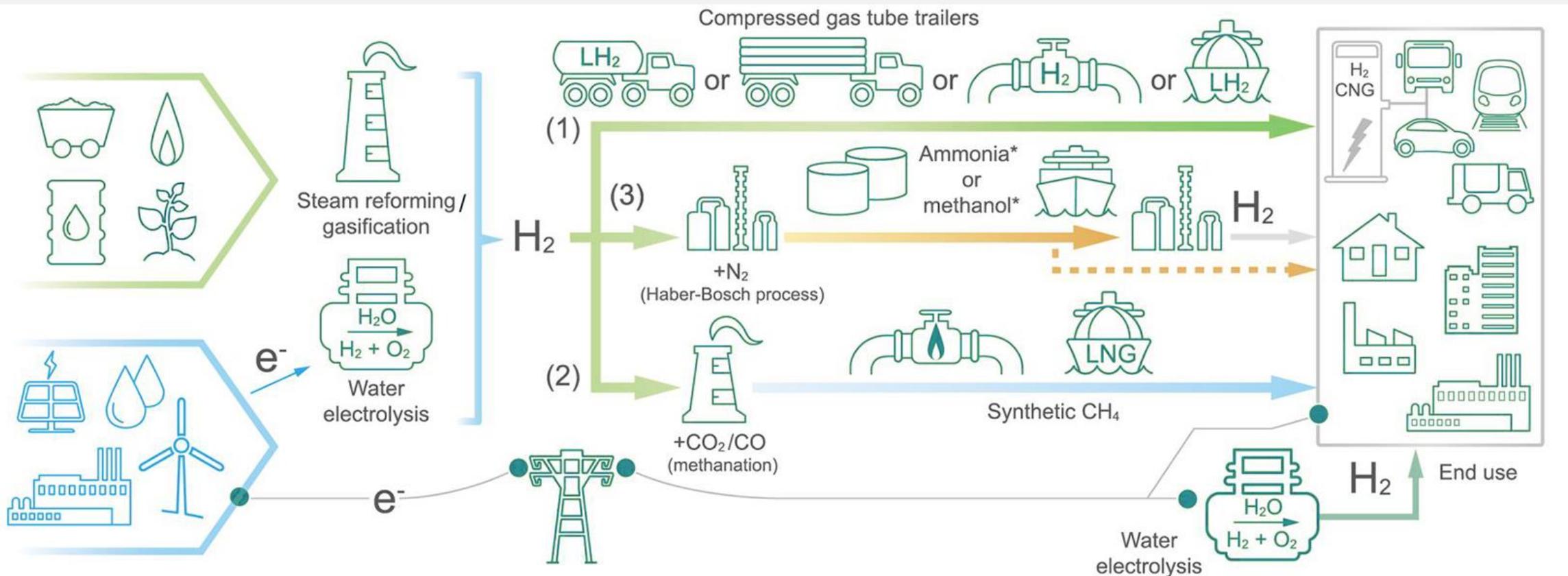
H₂ can play a key role in many economic sectors/uses

What is H₂ economy? | H₂ chain

Production

Storage/Transport & Distribution

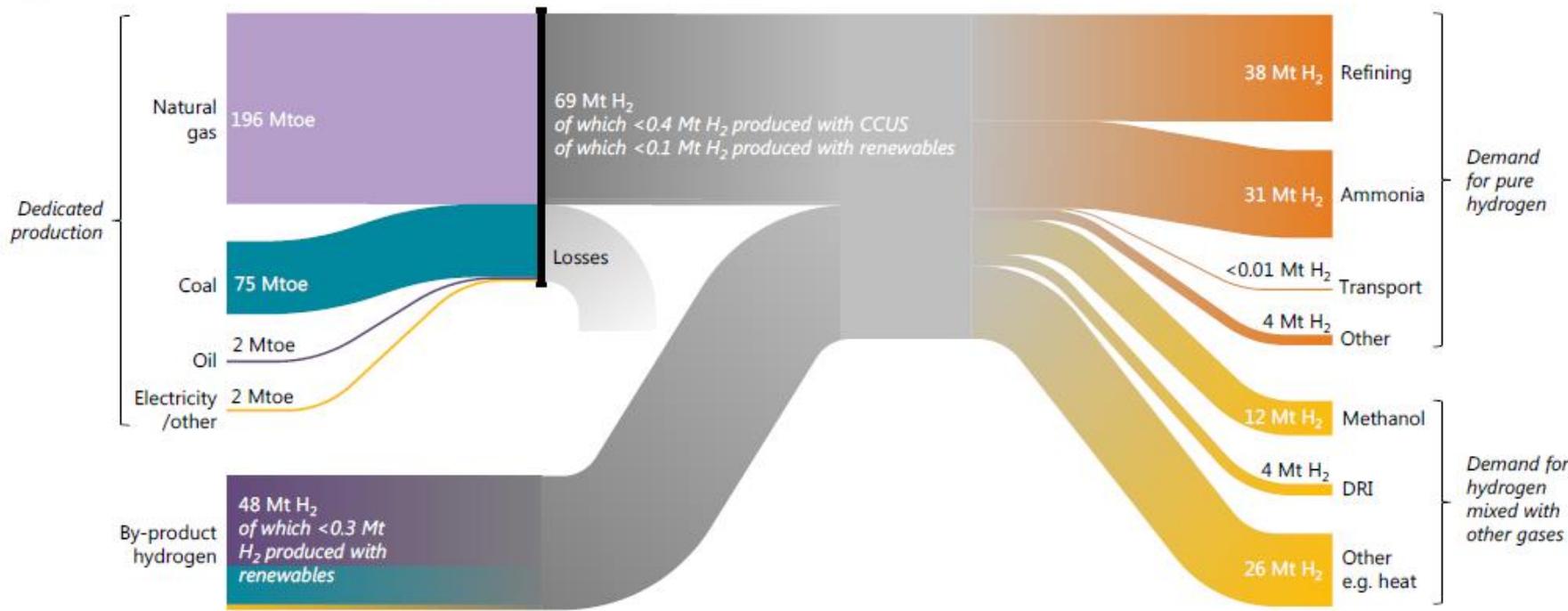
Consumption



Hydrogen value chains can follow many different paths

The production of H₂ today

Today's hydrogen value chains



IEA, 2019a

The majority of H₂ produced is from fossil fuels - 60% is from “dedicated” hydrogen production facilities

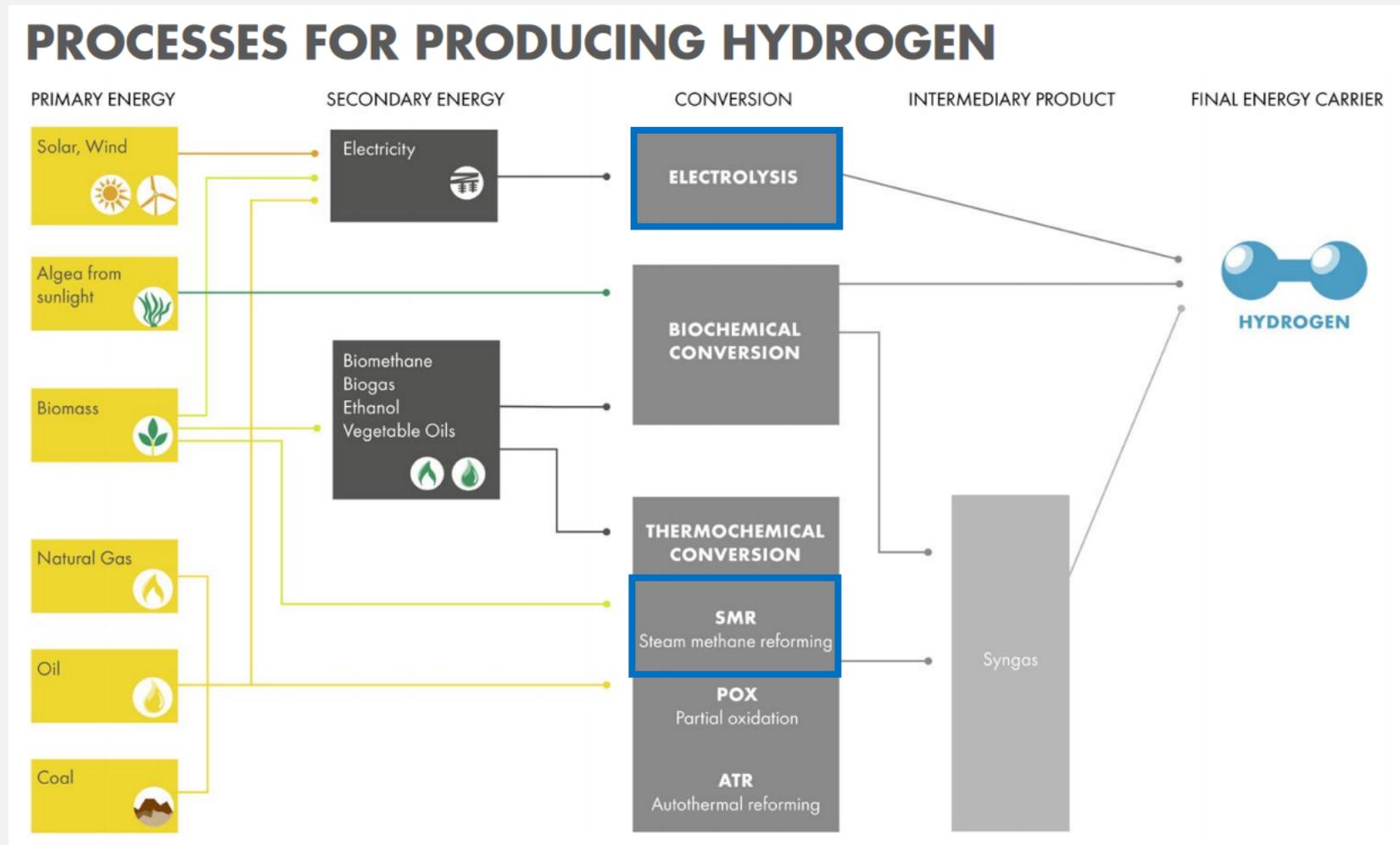
Most is produced from natural gas

Small fraction comes from water electrolysis (water + electricity).

Less than 0.7% of H₂ production is from RES or from fossil fuel plants with CCUS (2018)

H₂ production is responsible for 830 MtCO₂/yr

H₂ production



Grey/Blue (CCS) H₂

Black H₂

Source: Shell, 2017

Hydrogen 101 short film

HYDROGEN

3:23



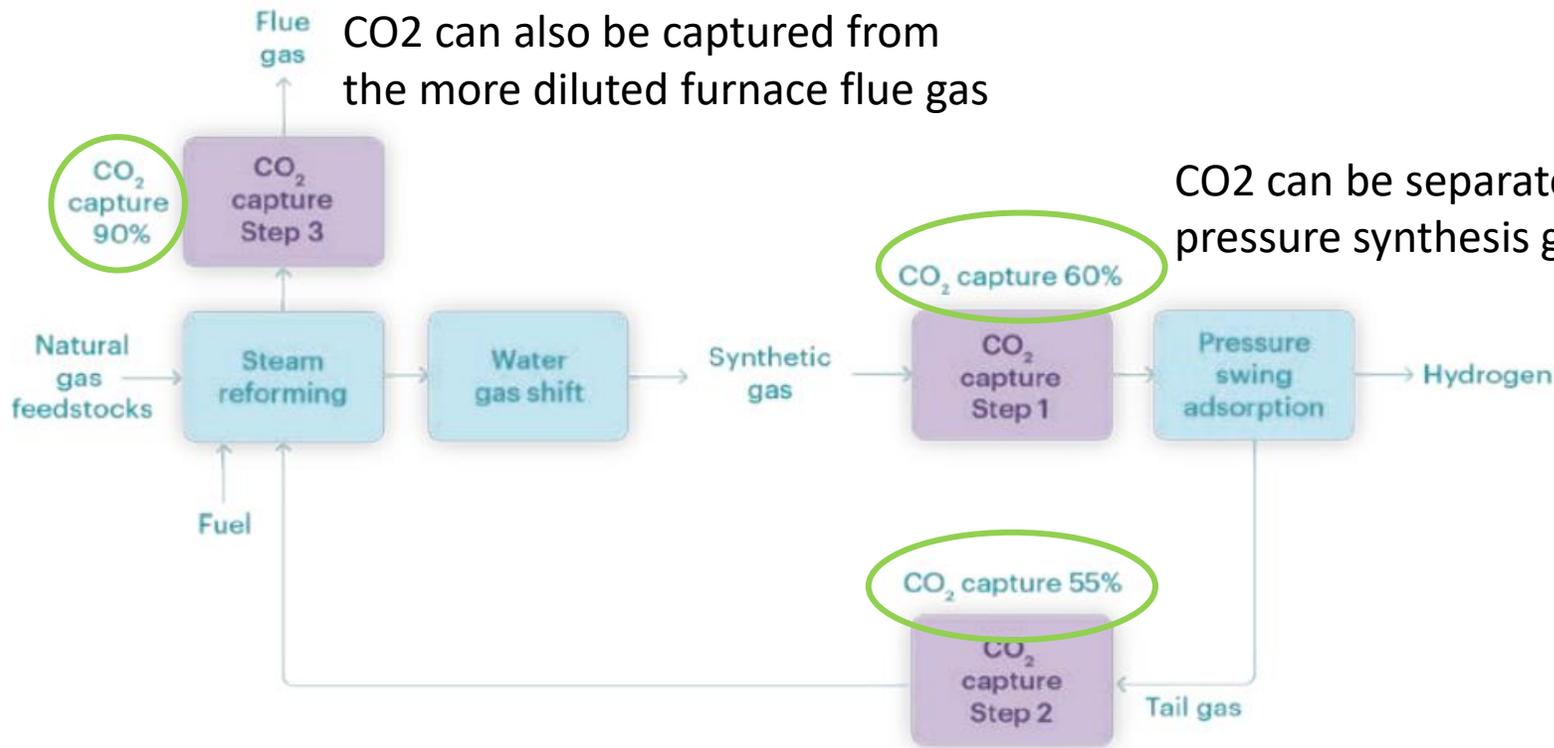
Student Energy
53,1 mil subscribers

<https://www.youtube.com/watch?v=Kv8WT3-7ZHE>

BLUE hydrogen

SRM with carbon capture and utilization

Figure 8. Production process of hydrogen from gas with CCUS



CO₂ can also be captured from the more diluted furnace flue gas

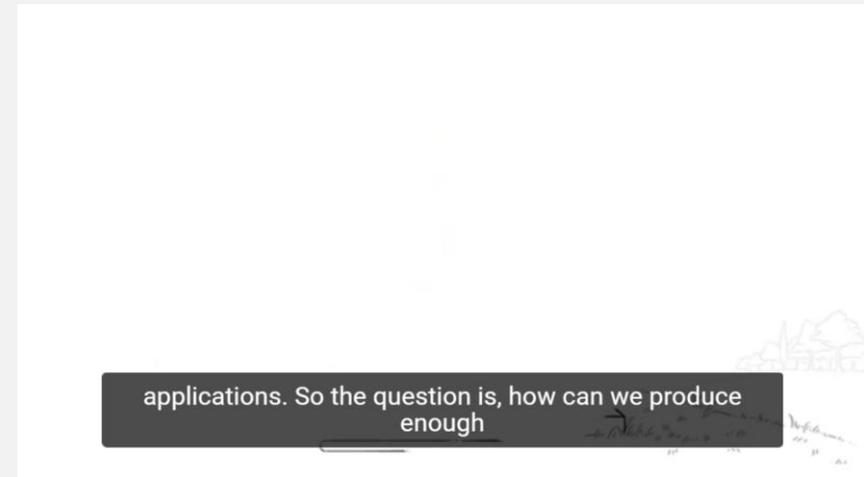
CO₂ can be separated from the high-pressure synthesis gas stream

Source: IEAGHG (2017a), "Reference data and supporting literature reviews for SMR based hydrogen production with CCS".

Source: IEA, 2019a

H₂ Production | Water electrolysis

- > Electrochemical reaction that splits water into H₂ and Oxygen, using electricity.
- > It is a 100% emission free and carbon-free process



<https://youtu.be/ZLf1cOWsefU>

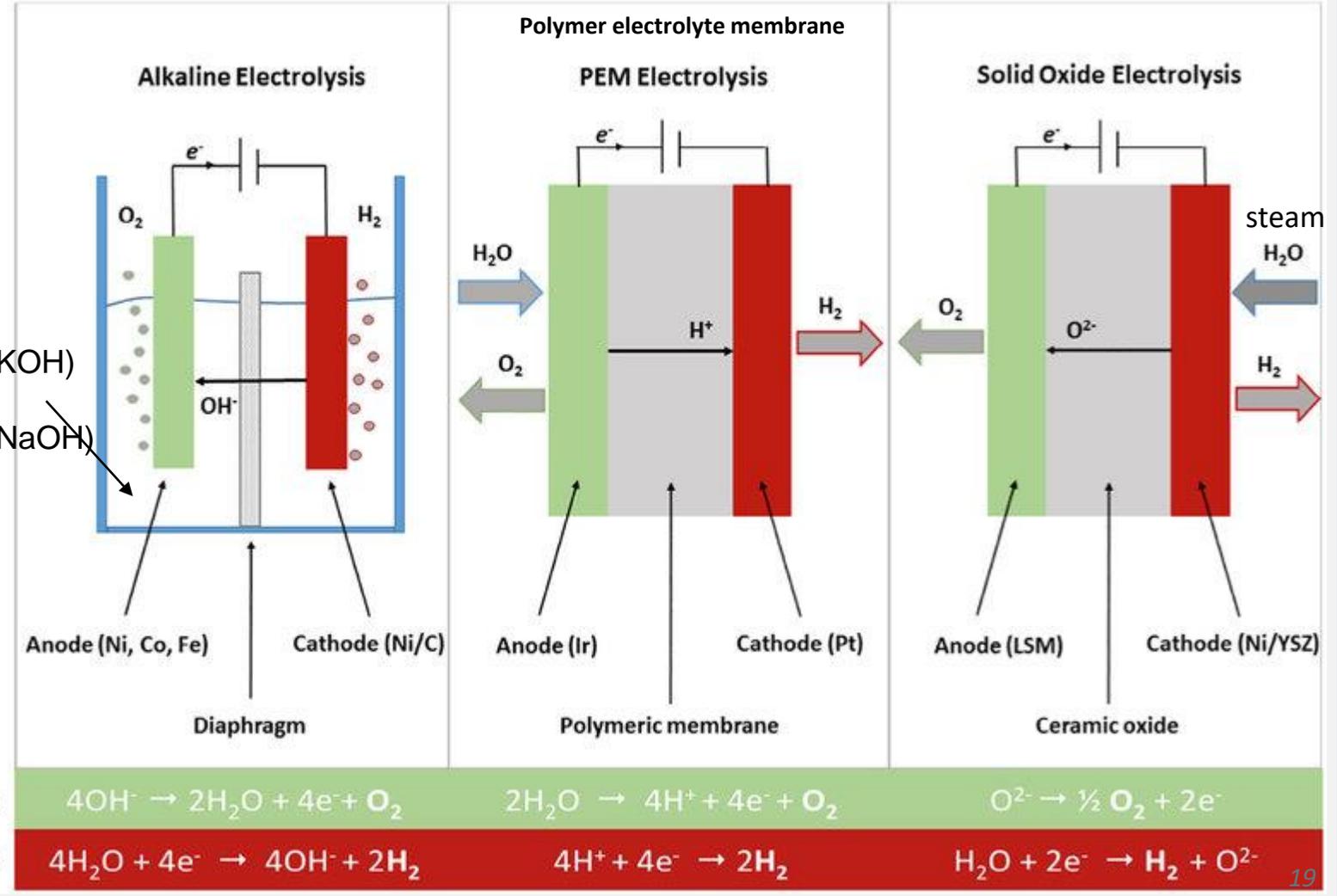
Power: 1 MW electrolyser ↔ ± 18 kg/h H₂

Energy: +/- 55 kWh of electricity → 1 kg H₂ ↔ ± 9 liters demineralized water

Water electrolysis

Electron to hydrogen

potassium hydroxide (KOH) or sodium hydroxide (NaOH) solution



Electron to hydrogen

Water electrolysis

> Alkaline electrolyser

- > Mature and commercial technology +
- > Do not operate on zero load -
- > Do not produce highly compressed H2 – needs additional compression -
- > Needs the recovery and recycling of the potassium hydroxide electrolyte solution -

> PEM

- > Produce highly compressed H2 +
- > Operating range can go from zero load +
- > Need expensive electrode catalysts (platinum, iridium) and membrane materials -

> SOEC

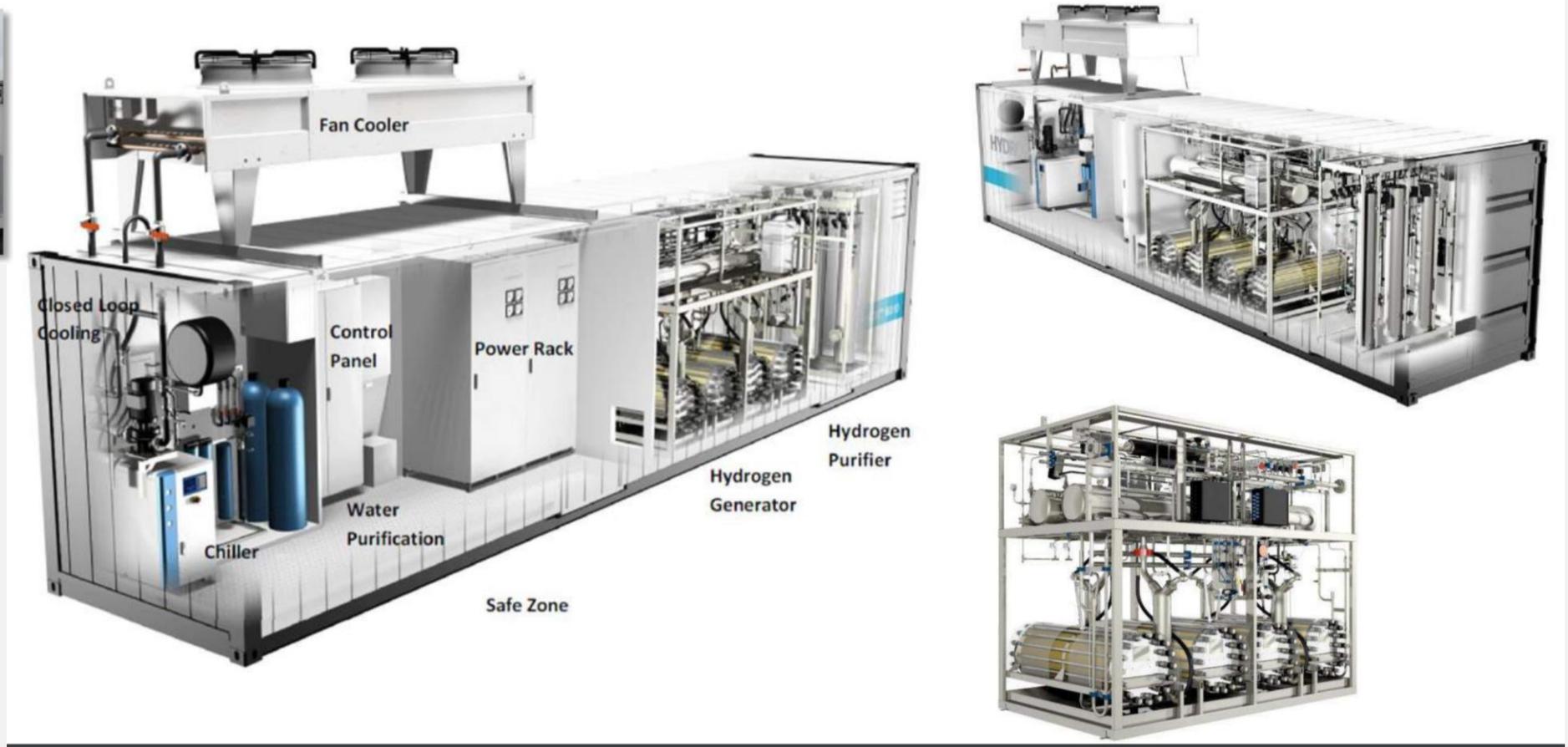
- > Have not yet been commercialized -
- > Need a heat source (nuclear, solar thermal, geothermal) -
- > It is possible to operate in reverse mode +

	Alkaline electrolyser			PEM electrolyser			SOEC electrolyser		
	Today	2030	Long term	Today	2030	Long-term	Today	2030	Long term
Electrical efficiency (% LHV)	63–70	65–71	70–80	56–60	63–68	67–74	74–81	77–84	77–90
Operating pressure (bar)	1–30			30–80			1		
Operating temperature (°C)	60–80			50–80			650 – 1 000		
Stack lifetime (operating hours)	60 000 – 90 000	90 000 – 100 000	100 000 – 150 000	30 000 – 90 000	60 000 – 90 000	100 000 – 150 000	10 000 – 30 000	40 000 – 60 000	75 000 – 100 000
Load range (% relative to nominal load)	10–110			0–160			20–100		
Plant footprint (m ² /kW _e)	0.095			0.048					
CAPEX (USD/kW _e)	500 – 1400	400 – 850	200 – 700	1 100 – 1 800	650 – 1 500	200 – 900	2 800 – 5 600	800 – 2 800	500 – 1 000

Studies indicate PEM as the main electrolyser technology in the future

Source: IEA, 2019a
20

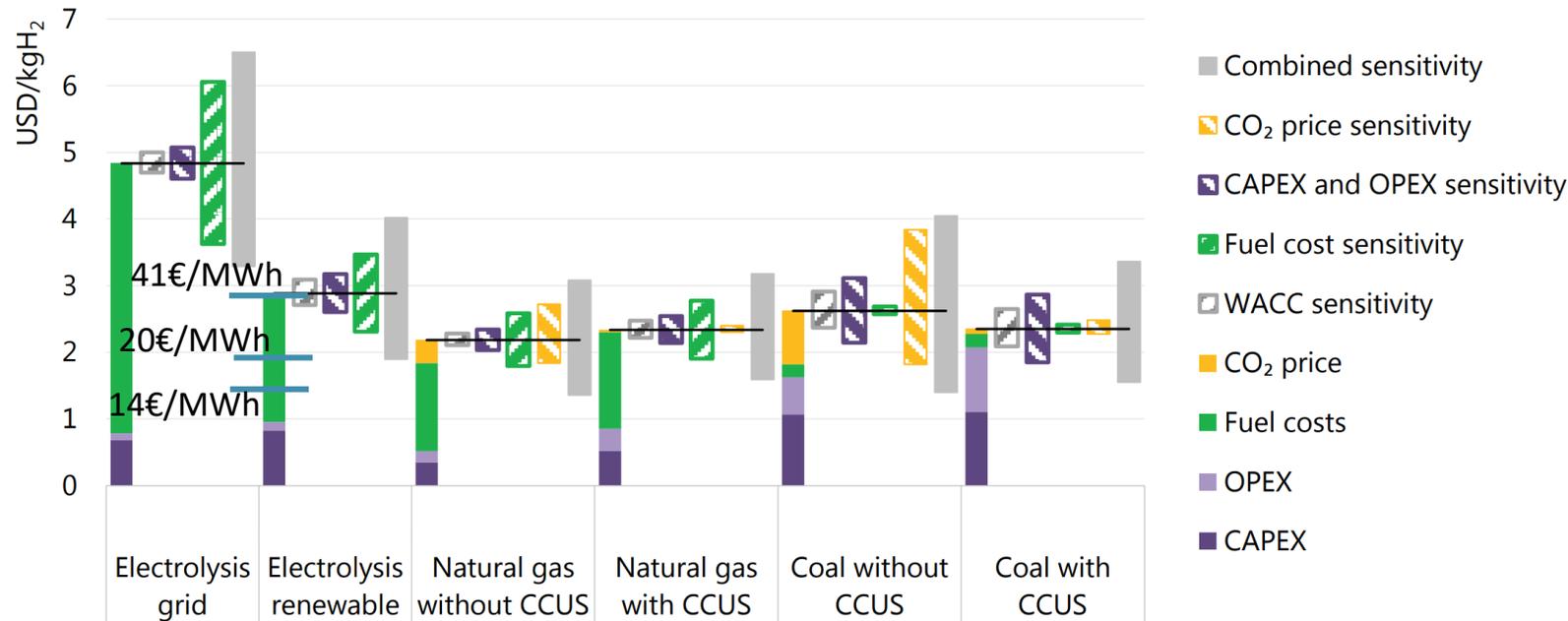
How does an electrolyser (alkaline) looks like?



Source:
https://hydrogeneurope.eu/sites/default/files/2018-06/2018-06_Hydrogenics_Company%20presentation.compressed.pdf

H₂ production costs

Hydrogen production costs for different technology options, 2030



Notes: WACC = weighted average cost of capital. Assumptions refer to Europe in 2030. Renewable electricity price = USD 40/MWh at 4 000 full load hours at best locations; sensitivity analysis based on +/-30% variation in CAPEX, OPEX and fuel costs; +/-3% change in default WACC of 8% and a variation in default CO₂ price of USD 40/tCO₂ to USD 0/tCO₂ and USD 100/tCO₂. More information on the underlying assumptions is available at www.iea.org/hydrogen2019.

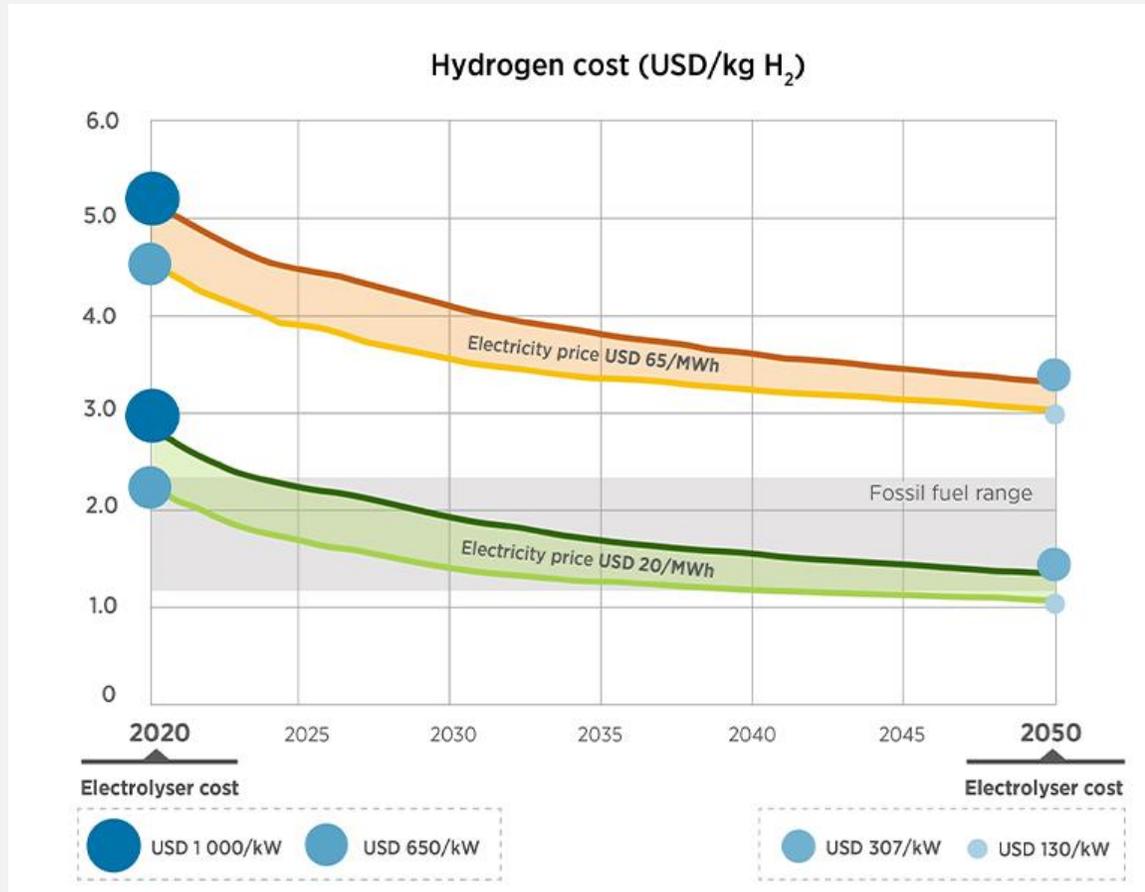
Source: IEA 2019. All rights reserved.

Source: IEA, 2019a

- > Steam Methane Reforming from Natural Gas is the cheapest way to produce H₂
- > Electricity price is the biggest component of H₂ production price (Renewable electricity price around 41€/MWh)
- > The 1st solar auction in Portugal has awarded at an average tariff of 18,4 €/MWh, with a lot awarded at 14,7 €/MWh, the lowest price in the world at the time. In the 2nd auction the lowest output price was broken once again with a bid of 11.2€/MWh

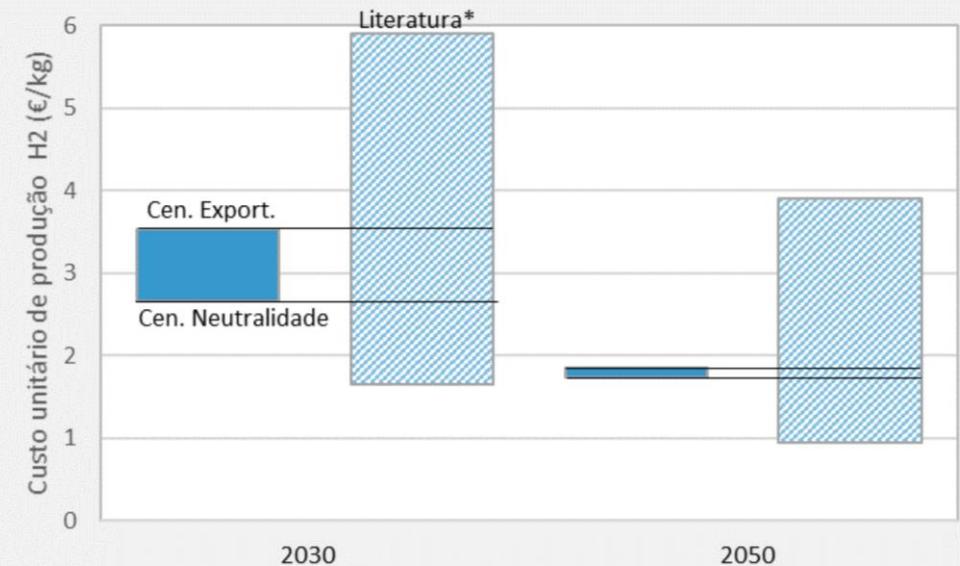
Electrolysis and H₂ cost evolution

- > Electrolysers are expected to reduce its cost between 65-80% from 2020 to 2050



Source: IRENA, 2020

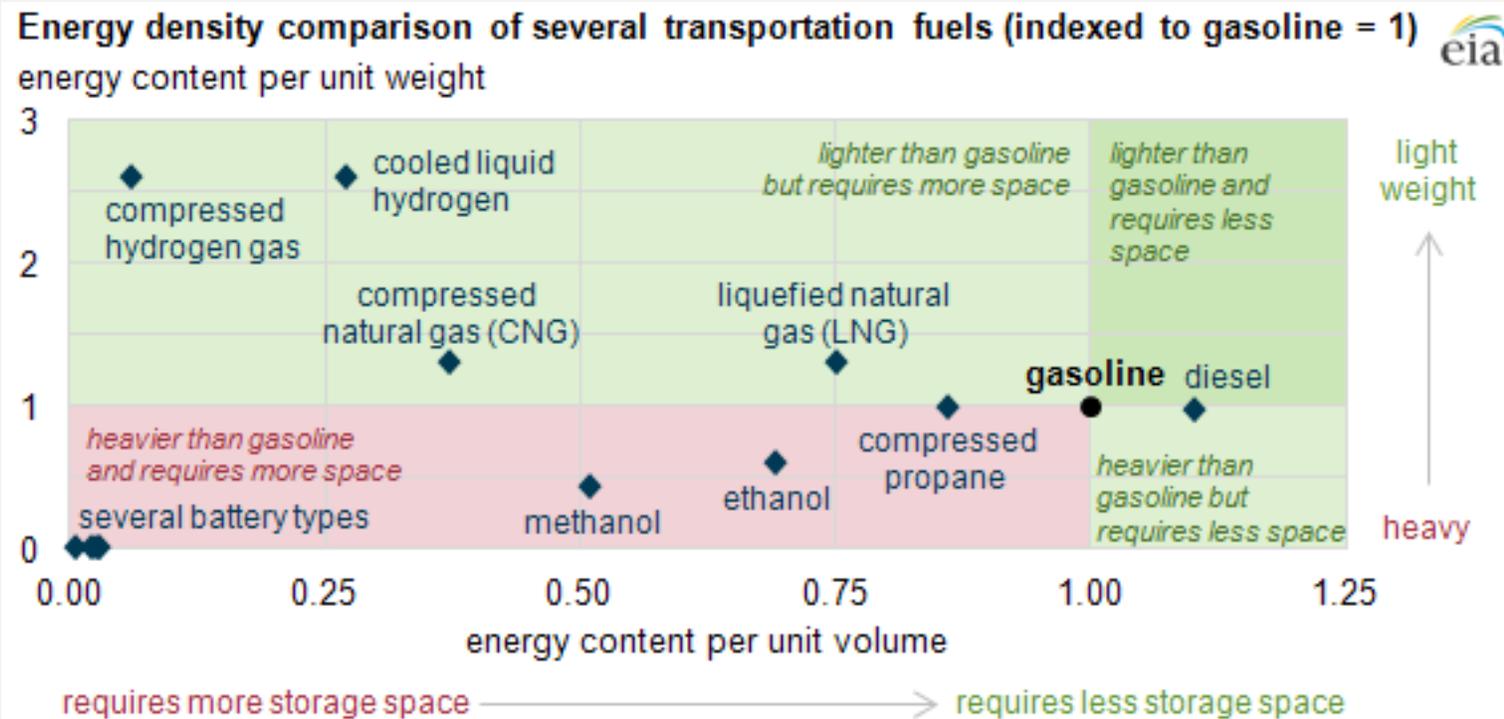
H₂ production cost in Portugal



Source: NOVA-FCT, 2020

H₂ Storage

- > H₂ low volumetric energy density at ambient conditions makes it considerably harder to store than fossil fuels – compressions, liquefaction or absorption
- > If hydrogen replace natural gas in the global economy today would need 3-4 times more storage infrastructure

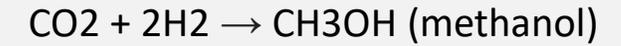
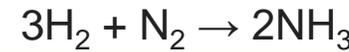


Source:
<https://www.eia.gov/todayinenergy/detail.php?id=9991>

H₂ Storage

Table 1: Hydrogen storage options

	Gaseous state				Liquid state			Solid state
	Salt caverns	Depleted gas fields	Rock caverns	Pressurized containers	Liquid hydrogen	Ammonia	LOHCs	Metal hydrides
Main usage (volume and cycling)	Large volumes, months-weeks	Large volumes, seasonal	Medium volumes, months-weeks	Small volumes, daily	Small - medium volumes, days-weeks	Large volumes, months-weeks	Large volumes, months-weeks	Small volumes, days-weeks
Benchmark LCOS (\$/kg) ¹	\$0.23	\$1.90	\$0.71	\$0.19	\$4.57	\$2.83	\$4.50	Not evaluated
Possible future LCOS ¹	\$0.11	\$1.07	\$0.23	\$0.17	\$0.95	\$0.87	\$1.86	Not evaluated
Geographical availability	Limited	Limited	Limited	Not limited	Not limited	Not limited	Not limited	Not limited



Cool it down H₂ further, to its melting point (-260°C)

- High energy lost (24-45%)
- Energy lost (5-10%)

Source: BloombergNEF. Note: ¹ Benchmark levelized cost of storage (LCOS) at the highest reasonable cycling rate (see detailed research for details). LOHC – liquid organic hydrogen carrier.

Source; BloombergNEF, 2020

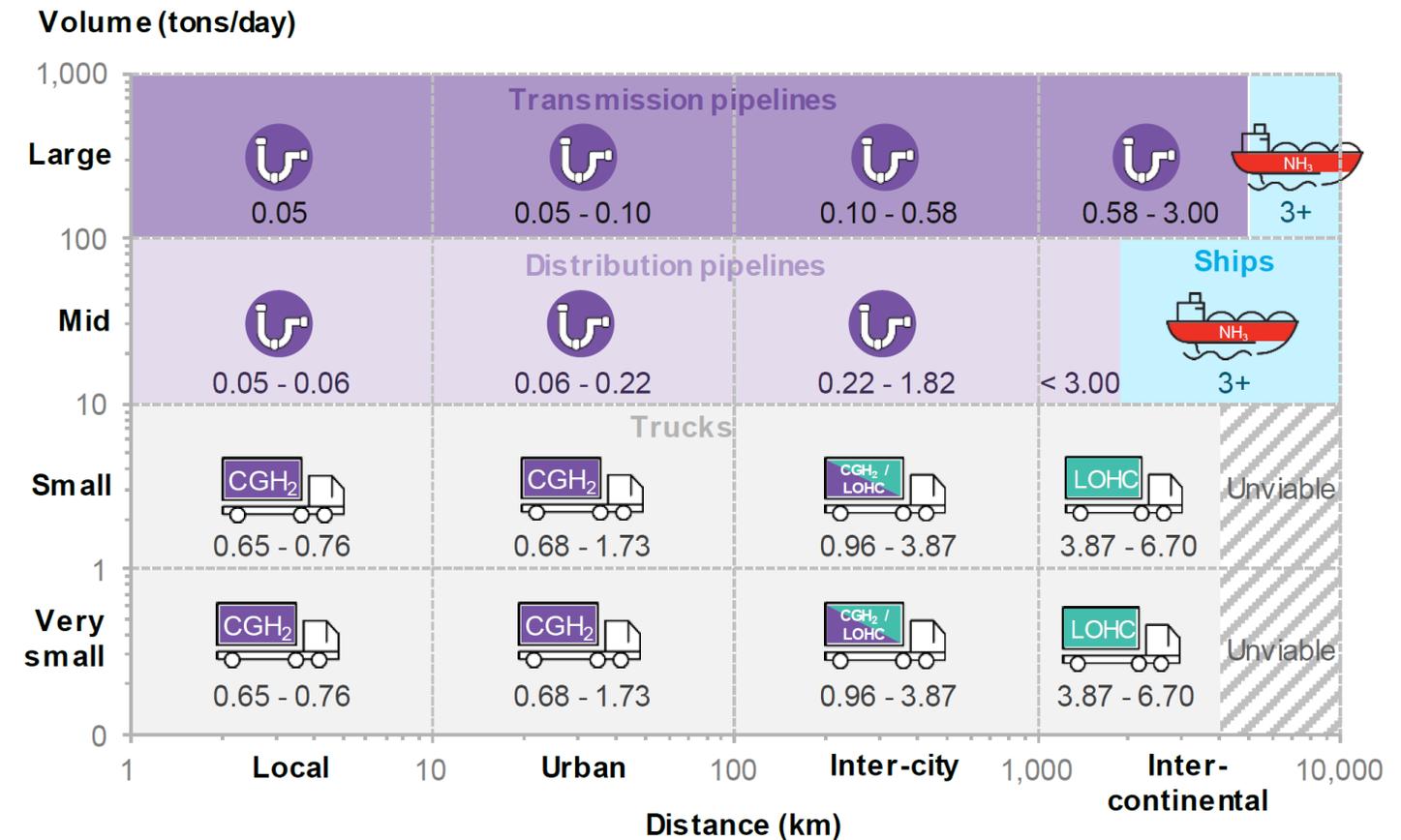
LOHC – Liquid Organic H₂ Carriers, e.g., methanol, dibenzyltoluene and toluene

Note: Salt caverns are the only type of geological formation successful used to storage H₂ underground to date. Other alternatives are under research to test leaks and reactivity with the host rock

H₂ Transport & Distribution

- > The low energy density of hydrogen means that it can be very expensive to transport over long distances
- > The best option: **blending in the natural gas grid, dedicated grid, trucks or shipping** will vary according to geography, distance, amount of H₂ and the required end use of the hydrogen

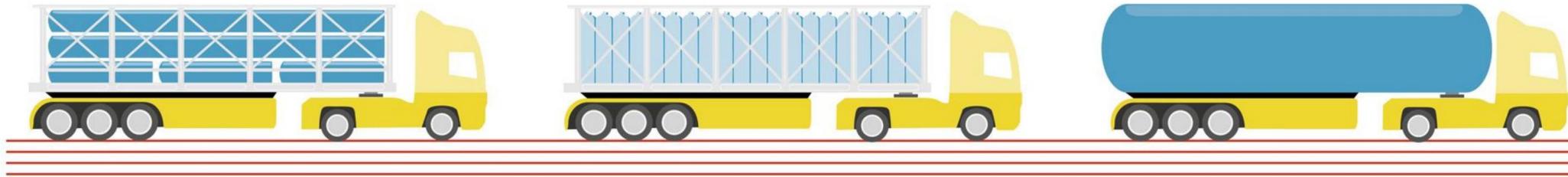
Figure 4: H₂ transport costs based on distance and volume, \$/kg, 2019



Source; BloombergNEF, 2020

Legend: Compressed H₂ Liquid H₂ Ammonia Liquid Organic Hydrogen Carriers

H₂ Road Transport



TUBE TRAILER

200 - 250 bar, ≈ 500 kg, ambient temperature

CONTAINER TRAILER

500 bar, ≈ 1,000 kg, ambient temperature

LIQUID TRAILER

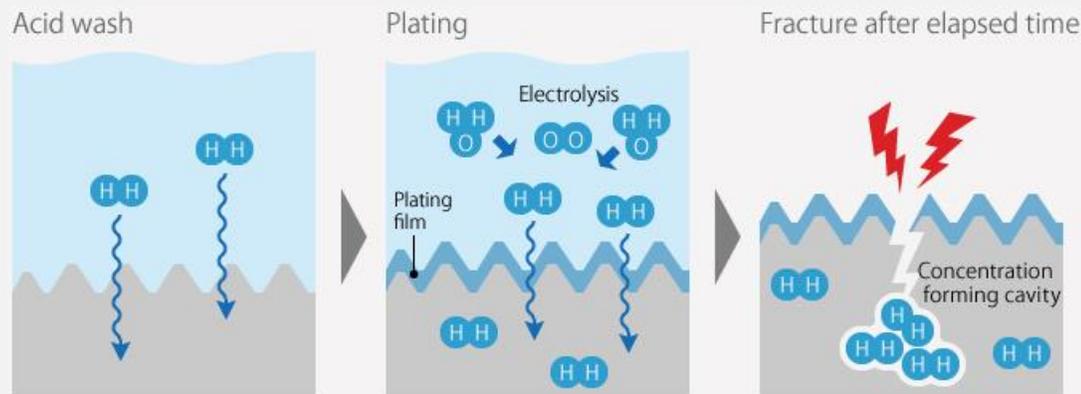
1 - 4 bar, ≈ 4,000 kg, cryogenic temperature

Source: Shell, 2017

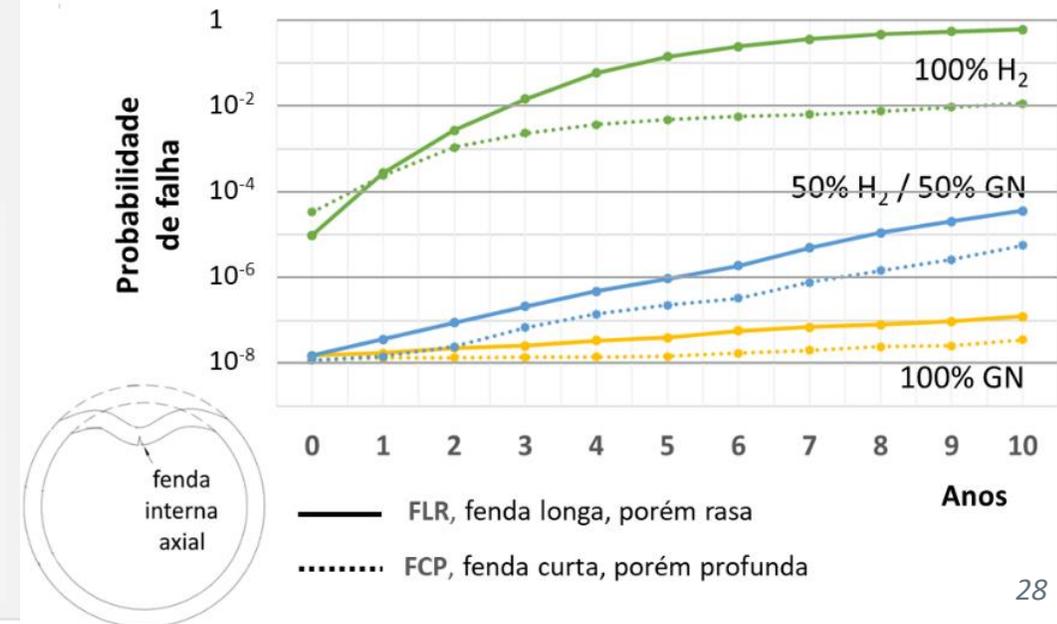
Blending H₂ in the natural gas grid

Limitations

- > The material of the pipeline limits the amounts of H₂:
 - > polyethylene distribution pipelines can handle up to 100% hydrogen
 - > some metal pipes can degrade when exposed to hydrogen over long periods, particularly with H₂ in high concentrations and at high pressures – **embrittlement** - Literature suggests a maximum of 20% blending without major transformations of the natural gas grid
- > Energy density of hydrogen is around 1/3 of that of natural gas and so a blend reduces the energy content of the delivered gas – more volume needed
- > *Even a 20% blending of H₂ only reduce natural gas emissions in 7%*

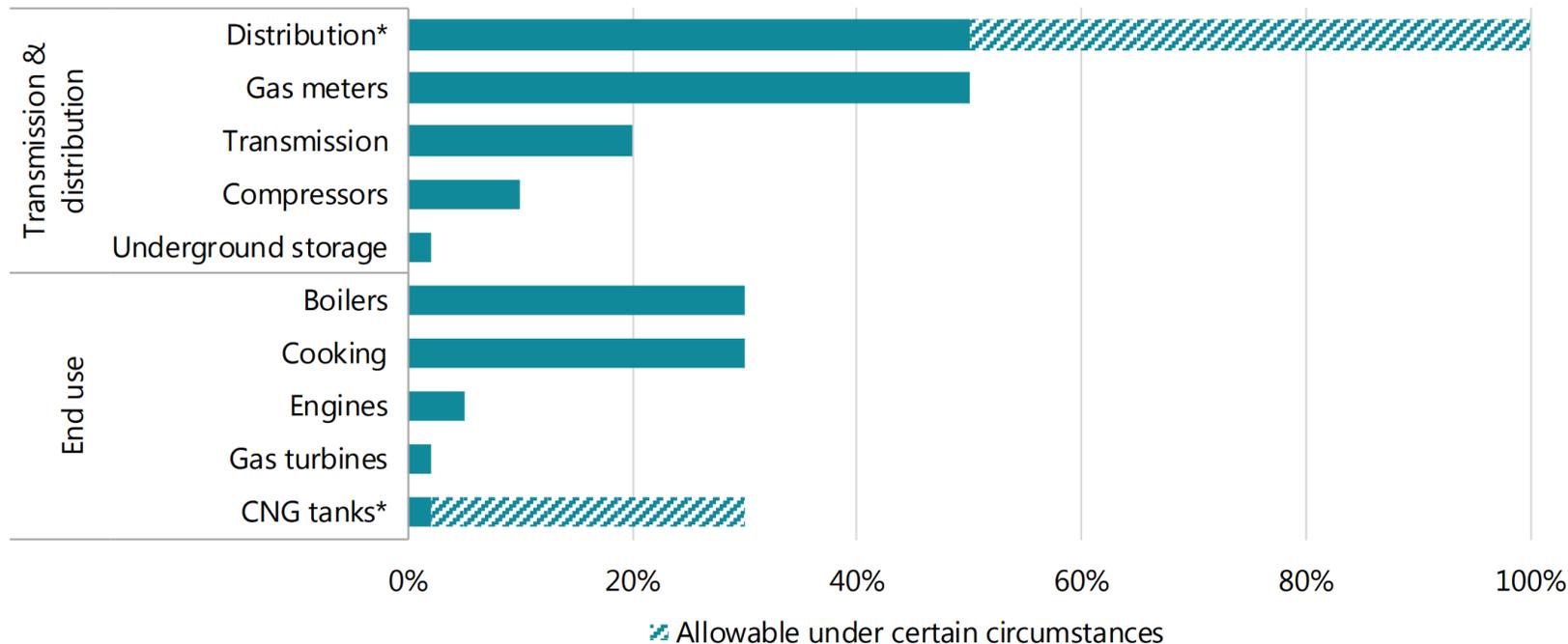


Fonte: Nabeya Bi-tech Kaisha (NBK®)



H₂ Blending in the natural gas grid

Tolerance of selected existing elements of the natural gas network to hydrogen blend shares by volume

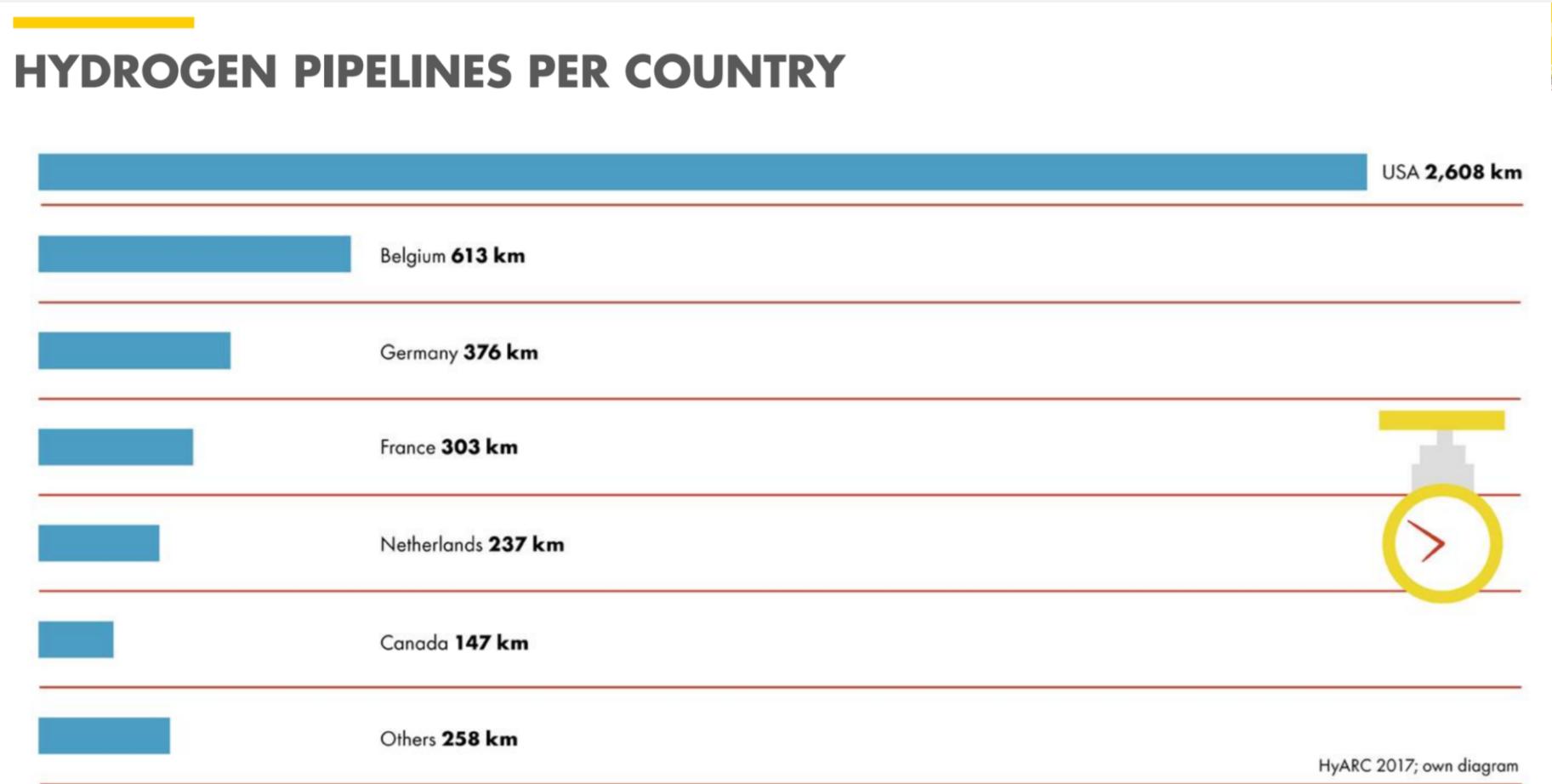


- > There are 37 demonstration projects studying H₂ blending in the gas grid.
- > The Ameland project in the Netherlands did not find that blending hydrogen up to 30% posed any difficulties for household devices, including boilers, gas hobs and cooking appliances

Source: IEA, 2019a

Dedicated H₂ pipelines

- > Dedicated H₂ pipeline already exist mostly associated with refineries/industry



Source: Shell, 2017

Green Pipeline Project (Portugal – Seixal)

- > Project leader: Galp Gás Natural Distribuição, Partners: PRF, Gestene, ISQ, Vulcano/Bosch
- > Starting: first half of 2020
- > Duration: 2 years
- > Budget: 0.5 M€

- > Goal: Mixing H₂ in the natural gas distribution network (polyethylene) starting with 2% and then increase to a maximum of 20% (v/v).
- > H₂ produced through Solar PV
- > 80 Consumers: 70 residential, one industrial and remaining tertiary

Overall H₂ economy

- > Production represents is the principal driver in H₂ costs
- > The higher the RES potential de lower the H₂ costs

Figure 5: Estimated delivered hydrogen costs to large-scale industrial users, 2030

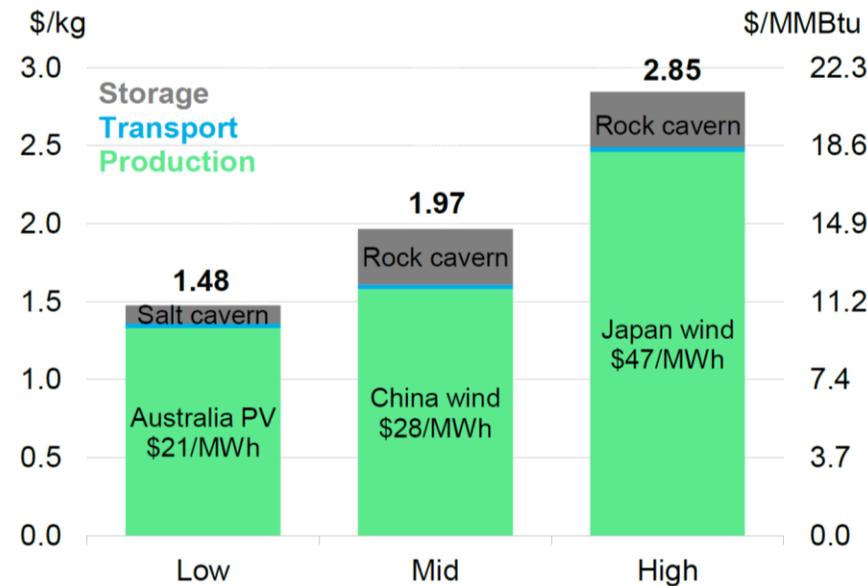
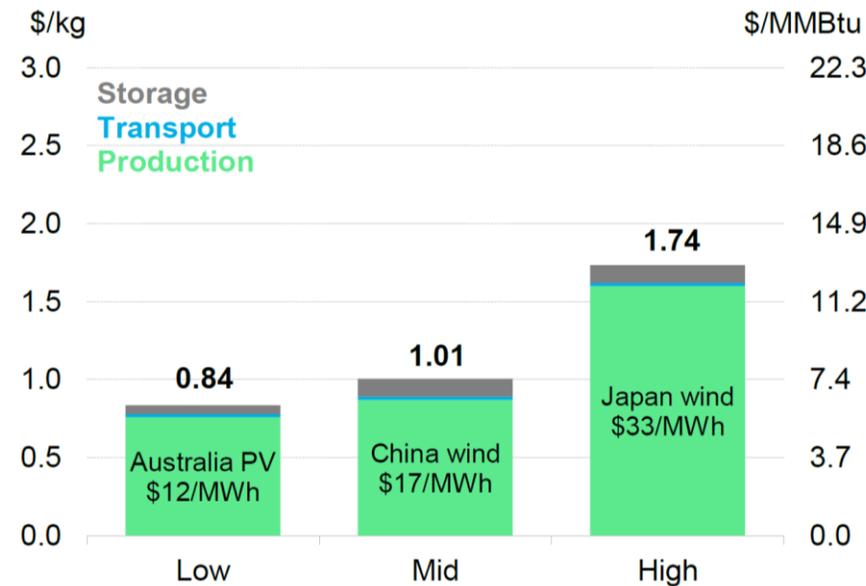


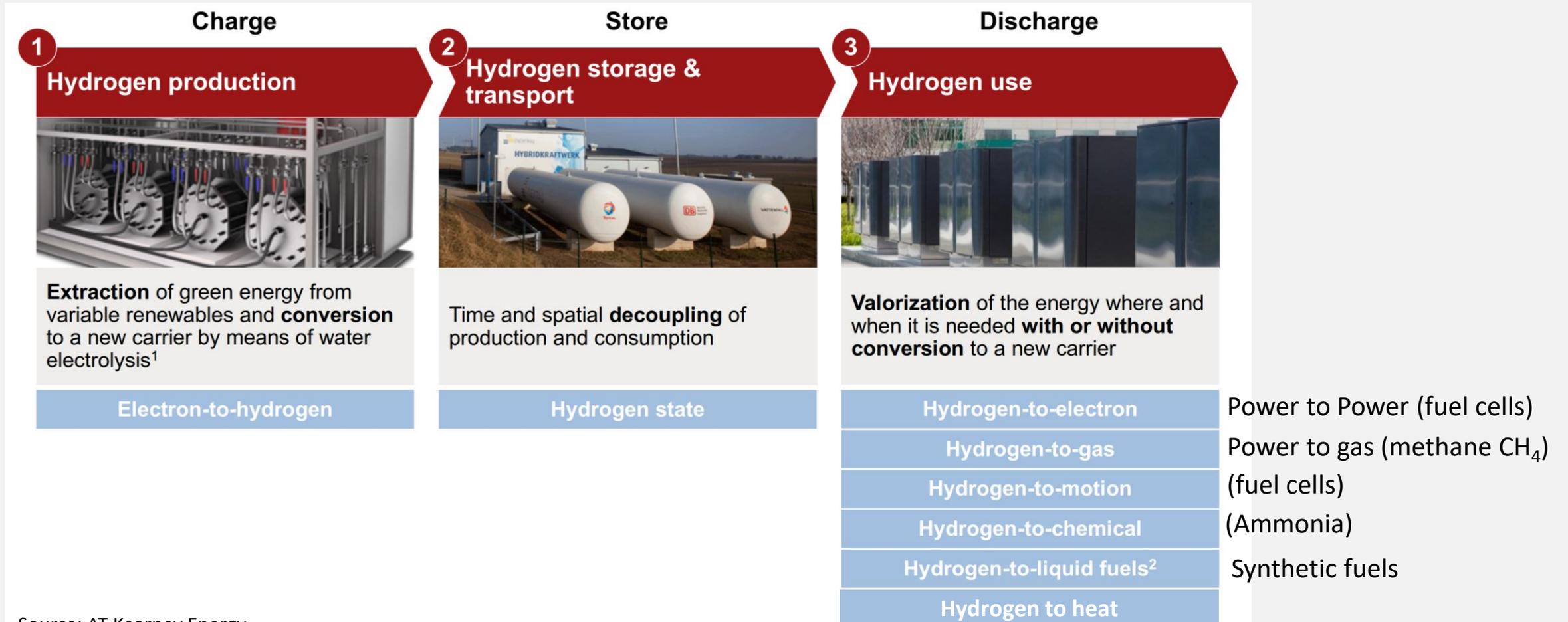
Figure 6: Estimated delivered hydrogen costs to large industrial users, 2050



Assuming the current costs of electricity from solar PV (last auction) Portugal will have a H₂ cost close to Australia

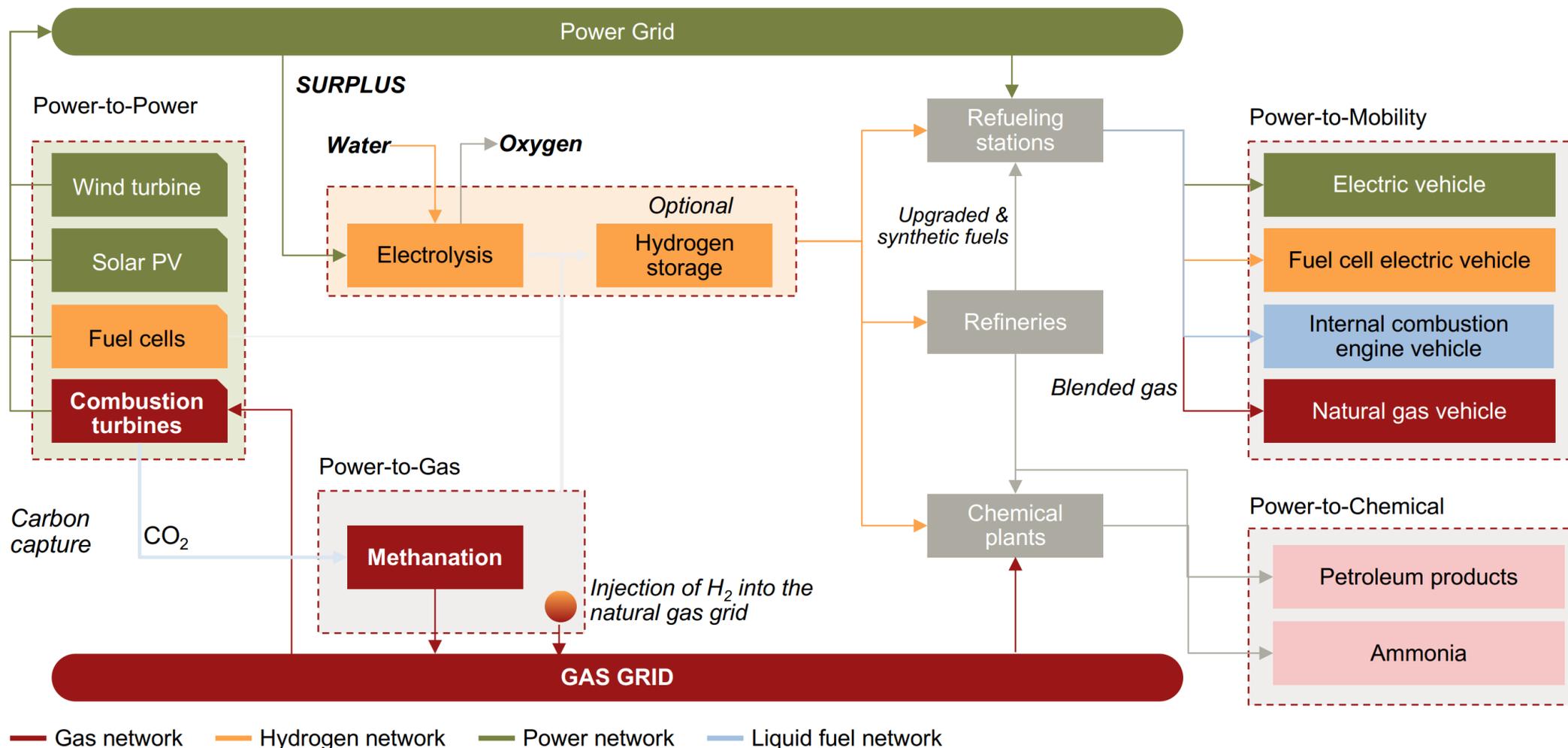
Transport costs represent 50km transmission pipeline

H₂ -based energy conversion solutions



Source: AT Kearney Energy Transition Institute, 2014.

H₂ -based energy conversion solutions

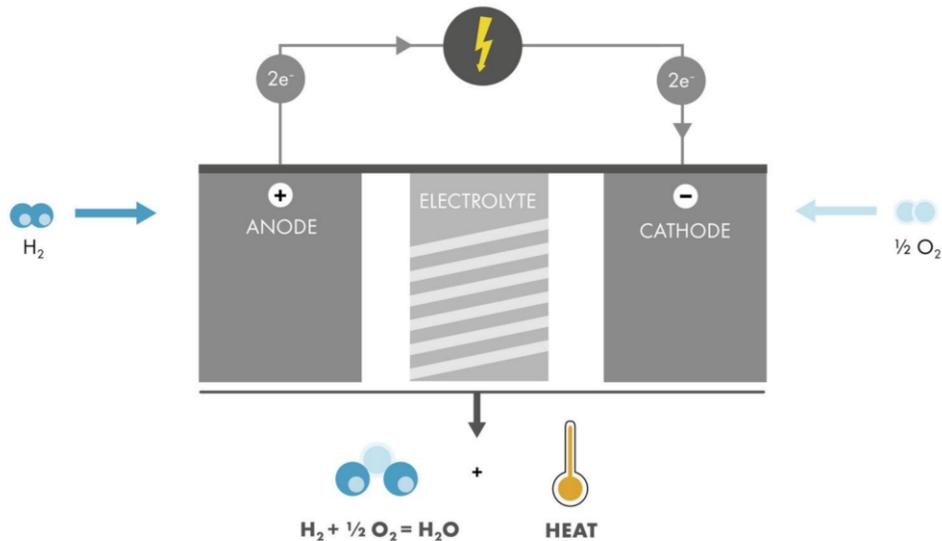


Source: AT Kearney Energy Transition Institute, 2014.

Fuel Cell

Power production from a hydrogen PEM fuel cell from hydrogen (+/- 50% efficiency)

Energy: 1 kg H₂ → 16 kWh



<https://www.youtube.com/watch?v=bXHwnKMchkk>

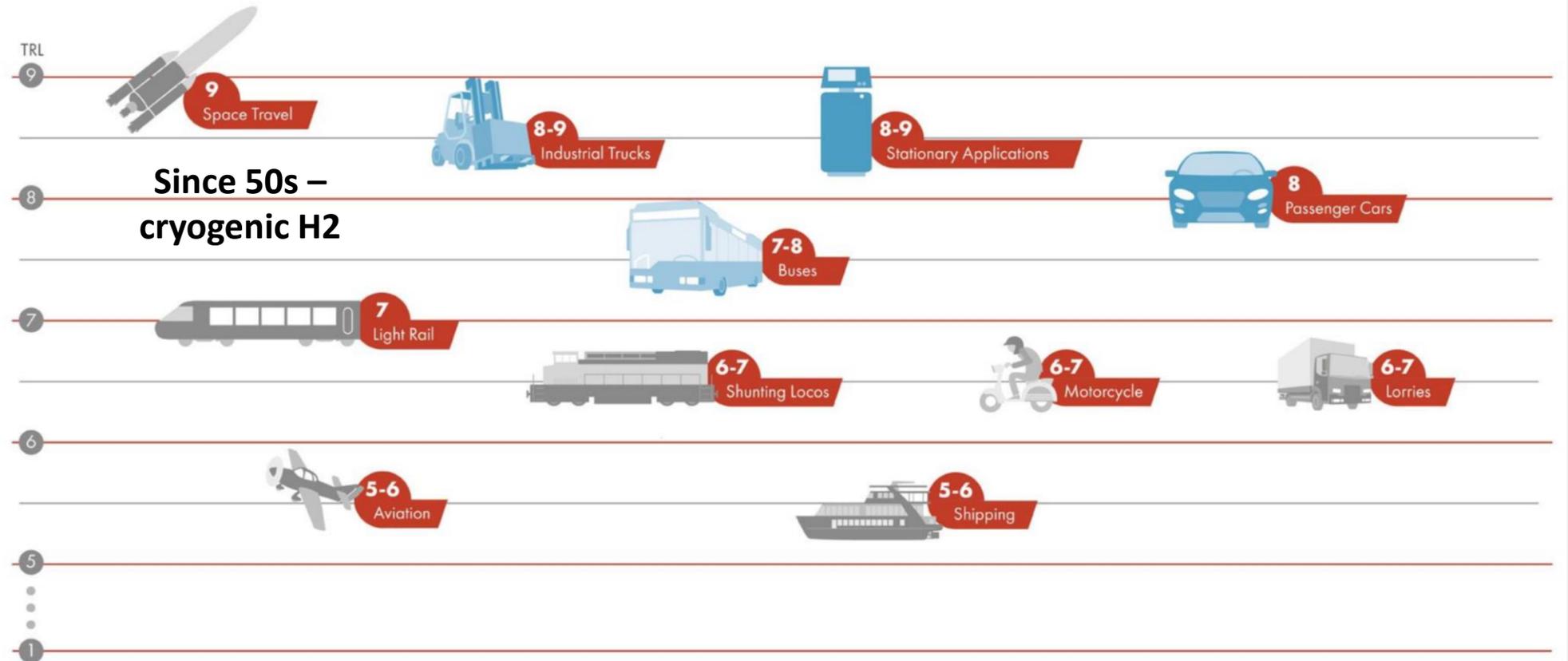
H₂ Uses – Transport Sector

TECHNOLOGY READINESS LEVELS OF HYDROGEN APPLICATIONS

System
complet and
quaified

Since 50s –
cryogenic H₂

Technology
validated in
relevant
environment



Source: Shell, 2017

Transport by H2 Fuel Cells – First Movers



- H₂ Fuel Cell bus trials began in many countries as long ago as the 1990's; substantial developments in 2000-2010.
- Feb, 2017: Toyota sells first FC bus to Tokyo Metropolitan Govt.; > 100 buses by 2020 Olympics*.

*<https://global.toyota/en/detail/15160167>



- 100km track between Cuxhaven and Buxtehude, Nth Germany; the train runs for 1,000km on one tank of H₂.
- France, Germany, Holland, Scandanavia and United Kingdom are first-movers with FC trains.
- JR East, Japan plans to test fuel cell trains in 2021.

- Nicola Motors aims at 700 refuelling truck stops in USA by 2028
- Claims 12-15 mpg compared with 6 mpg for diesel
- Anglo-American partner with Williams Engineering to build hydrogen powered ultra-class electric mining haul truck.

<https://energypost.eu/hydrogen-fuel-cell-trucks-can-decarbonise-heavy-transport/>
Australian Minina. Februarv 2020

Source: <http://www.dsdmip.qld.gov.au/resources/presentations/cq-hydrogen-presentation-2.pdf>

Fuel cell electric vehicle (FCEV) in Portugal

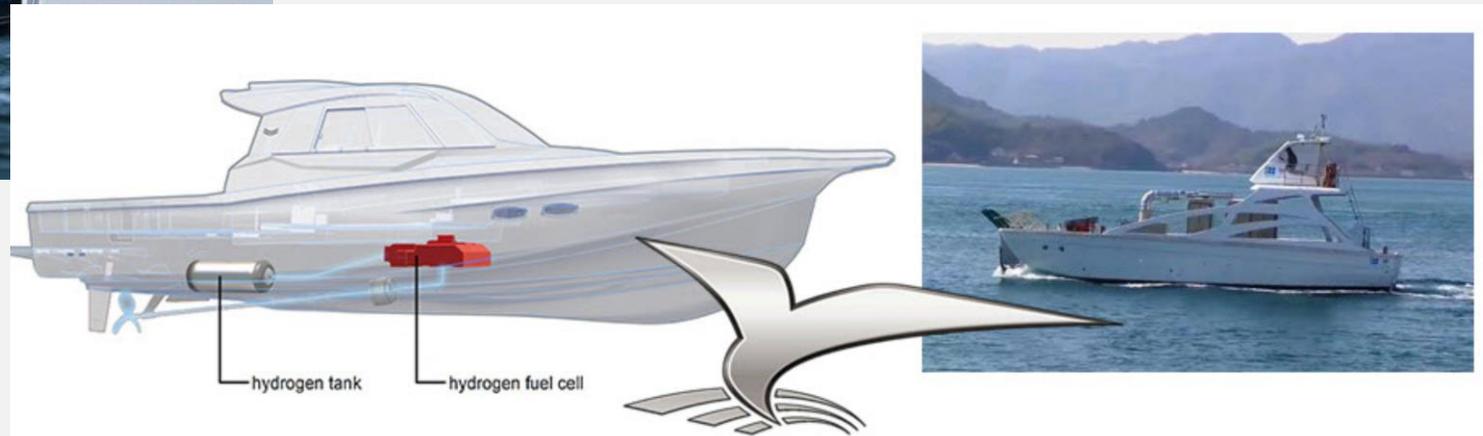
- > Caetano Bus (*fuel cell* Toyota) manufacture the H2.City Gold



Navigation



Source: <https://www.energy-reporters.com/>



Source: <https://fuelcellsworks.com/>

YANMAR

H₂ in aviation (2035)

Introducing Airbus ZEROe

Turboprop



<100
Passengers



Hydrogen
Hybrid Turboprop
Engines (x 2)



1,000+nm
Range



Liquid Hydrogen
Storage & Distribution
System

Blended-Wing Body



<200
Passengers



Hydrogen
Hybrid Turbofan
Engines (x 2)



2,000+nm
Range



Liquid Hydrogen
Storage & Distribution
System

Turbofan



AIRBUS

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